Theme	Elements of Life	Developing Fuels	Linder the Sea	The Ozone Story	Soans Scents	The Analysis
meme					and Flavours	
Specification	1.1 Atomic	2.1 Introduction	2.2 Croup 7	2.2 Alkanos		2 6 Organic
Specification		5.1 Introduction	2.5 Group /	3.2 Alkalles	3.5 AICOHOIS	5.0 Urganic
Points	Structure	to Organic	1.7 Oxidation,	3.3	3.6 Organic	
	1.2 Amount of	Chemistry	Reduction and	Halogenoalkanes	Analysis	3.15 NIVIR
	Substance	3.2 Alkanes	REDUX	1.5 Kinetics	3.9 Carboxylic	
	1.3 Bonding	1.4 Energetics	equations	1.3 Bonding	Acids and their	
	2.2 Periodicity	1.2 Amount of	1.6 Equilibria		derivatives	
		Substance				
		3.4 Alkenes				
Key	What is in our	Where does the	How can	Why is there are	How are soaps	How can we
Questions	blood, sweat and	energy come	bromine be	hole in the ozone	and detergents	determine the
	seas? What do	from? How	extracted from	layer?	made?	structural
	the molecules	versatile are	sea water?	Why do some	How can we	formula of a
	'look' like?	alkenes?		gases contribute	distinguish	compound?
				to the	between	
				greenhouse	colourless	
				effect?	organic	
					compounds?	
equired	1a: Preparation	2: Enthalpy	4b: Tests for	3: Effect of	5: Distillation of	
ractical*	of a standard	change	anions	temperature on	an organic	
	solution			the rate of	product	
	1b: Acid-base			reaction	6: Tests for	
	titration				organic	
	4a: Tests for				functional	
	cations				groups	
					10b:	
					Preparation of	
					an organic liquid	

<u>Course Overview</u>: Over the course of your 2 years here we will be studying chemistry through the following 12 themes

At the end of year 1 as well as mock revision you will also carry out some **practical investigations** such as investigating the vitamin C content of different fruits and vegetables, and extracting limonene from citrus fruits.

	Theme	Pharmaceutical	The Chemical	Amazing Metals	Oceans	Polymers and	The Synthesis
		Chemistry	Industry			Life	Puzzle
	Specification	3.10 Aromatic	1.9 Rate	2.5 Transition	1.8	3.12 Polymers	3.14 Organic
	Points	Chemistry	Equations	Metals	Thermodynamics	3.7 Optical	Synthesis
		3.11 Amines	1.10 The	1.11 Electrode	2.4 Period 3	Isomerism	
		3.9 Carboxylic	Equilibrium	Potentials and	elements and	3.13 Amino	
		Acids and their	Constant K _p	Electrochemical	their oxides	Acids, proteins,	
		derivatives		Cells	1.12 Acids and	and DNA	
		3.8 Aldehyes		2.6 Reactions of	Bases	3.16	
		and Ketones		ions in aqueous		Chromatography	
		3.7 Optical		solution			
		Isomerism		3.7 Optical			
				Isomerism			
	Кеу	How were the	How do rate	Why do	What is the role	How is nylon	How are organic
	Questions	modern ideas	equations help	transition metals	of oceans in	produced?	compounds
		about medicine	us understand	have coloured	climate control?	What are	synthesised?
		developed?	mechanisms?	compounds?		proteins?	
	Required	10a: Preparation	7: Rate of	8: EMF of	9: pH curves	12: Thin Layer	
	Practical*	of an organic	reaction by a)	electrochemical		chromatography	
		solid	initial rates	cell			
			method b) Rate	11: Transition			
2			of reaction by	metal ions			
ear			continuous				
¥			monitoring				